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marfulo mooctyliap nocii nonia odbn
starscly tonica Clues Vocabulary Terms
A type of bond that holds cations and anions together
A type of bond in which two atoms share a pair of valence

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electrons

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Chapter 6 Chemical Bonds Summary 6.1
Ionic Bonding When the highest occupied energy level of an atom is filled with electrons, the atom is stable and not likely to react. • The chemical properties of an element depend on the number of valence electrons.

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Chapter 6 Chemical Bonds - Henry County School District

CHAPTER 6 REVIEW Chemical Bonding
SECTION 5 SHORT ANSWER Answer the
following questions in the space
provided. 1. Identify the major
assumption of the VSEPR theory, which
is used to predict the shape of atoms.
Pairs of valence electrons repel one

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another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

6 Chemical Bonding

Writing Formulas for Ionic Compounds

What is the ratio of the ions in magnesium iodide? What is the formula for magnesium iodide? 1. Read and

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Chapter 6 -Chemical Bonds | Covalent Bond | Ion

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Level B Chapter 6 65 IPLS Section 6.4
The Structure of Metals (pages 176–181)
This section discusses metallic bonds
and the properties of metals. It also
explains how the properties of an alloy

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are controlled. <http://mrmicco.weebly.com/uploads/1/2/5/8/12584959/psgs1064b.pdf>

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