

## Class Xii Chemistry Ch 2 Solutions

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### Class Xii Chemistry Ch 2

Henry's law constant for CO<sub>2</sub> in water is  $1.67 \times 10^8$  Pa at 298 K. Calculate the quantity of CO<sub>2</sub> in 500 mL of soda water when packed under 2.5 atm CO<sub>2</sub> pressure at 298 K. Ans.: 2.8 The vapour pressures of pure liquids A and B are 450 mm and 700 mm of Hg respectively at 350 K. Calculate the composition of the liquid mixture if total vapour ...

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All About The Chapter 2 Chemistry Class 12 Solution. Chapter 2 of NCERT Chemistry is based on the basic idea of Solutions and its types. This chapter covers Concentration of Solutions, Solubility, Vapour Pressure of Liquid Solutions, Ideal and Non-Ideal Solutions, Colligative Properties and Determination and Abnormal Molar Masses.

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Chemistry Class 12 solutions Chapter 2 can be utilised by the students to prepare for the examination and to solve the questions of class 12 chemistry chapter 2 exercise solutions along with exemplar problems, MCQS, short and long answer questions. These solutions help the students to clear all their doubts related to this chapter very easily.

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Ans: Following are some of the important questions students must practice from Class 12 Chemistry Chapter 2: Give an example of 'liquid in solid' type solutions. Under What Condition is Van't Hoff Factor less than 1? What is reverse osmosis? Give one large scale use of it? Define cryoscopic

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constant (Molal Freezing Point Depression Constant).

### **Important Questions for CBSE Class 12 Chemistry Chapter 2 ...**

Important Questions for Class 12 Chemistry Solutions Chapter 2 will help the students for board and competitive exams. Understand solutions of Mole Fraction, Henry's Law Constant, Vapor Pressure and more.

### **Important Questions for Class 12 Chemistry Solutions Chapter 2**

Important Questions for Class 12 Chemistry Chapter 2 Solutions Class 12 Important Questions Solutions Class 12 Important Questions Very Short Answer Type Question 1. Differentiate between molarity and molality of a solution. (All India 2010) Answer: The distinction between molarity and molality. Molarity : It is the number of moles of solute dissolved in 1 [...]

### **Important Questions for Class 12 Chemistry Chapter 2 ...**

Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

### **Chemistry Notes for class 12 Chapter 2 Solutions**

NCERT Books Class 12 Chemistry: The National Council of Educational Research and Training (NCERT) publishes Chemistry textbooks for Class 12. The NCERT Class 12th Chemistry textbooks are well known for it's updated and thoroughly revised syllabus. The NCERT Chemistry Books are based on the latest exam pattern and CBSE syllabus.

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Class 12 Chemistry Chapter 2 Solutions The second chapter deals with types of solutions, and their properties. The concepts dealt with are Raoult's law, concentration of solutions, vapor pressure of liquid solutions, abnormal molar masses, and colligative properties.

### **NCERT Solutions for Class 12 Chemistry (Updated for 2020-21)**

In NCERT PDF Class 12 chemistry, concepts like crystalline solids, unit cell dimension and lattice structures have been discussed in detail and emphasized on. Chapter 2: Solutions. Another important class 12 chemistry chapter is solutions. A solution is defined as a homogenous blend which is made by mixing two or more solutes and solvents.

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NCERT Solutions for Class 12 Chemistry Chapter 2 - Solutions. This article deals with NCERT Solutions for Class 12 Chemistry Chapter 2. Chemistry refers to the study of matter and its properties. This along with various formulae makes Chemistry a difficult subject. However, NCERT Solutions removes the fear of this subject.

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Thus, CBSE Class 12 Chemistry important questions would help them to prepare for both kinds of examinations. Students can practice on our chapter-wise Important questions for Class 12 Chemistry which will enable them in developing conceptual knowledge. Important Questions for Class 12 Chemistry

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Class XII Chemistry Chapter 2 Solutions NCERT Solution is given below. Suggest the most important type of intermolecular attractive interaction in the following pairs. (v) acetonitrile ( $\text{CH}_3\text{CN}$ ) and acetone ( $\text{C}_3\text{H}_6\text{O}$ ). (i) Van der Waals forces of attraction. (ii) Van der Waals forces of attraction.

### **NCERT Solutions for Class 12 Chemistry Chapter 2 Solutions ...**

HSSLIVE Chemistry Plus Two Notes - Kerala Board Class 12 Chemistry - Here is a detailed Notes, Syllabus, Explanation, Summary, Difficult words, Question Answers provided for HSSLIVE Chemistry Class 12. Every plus two Chemistry chapter wise notes here been provided with complete explanations from beginning to the end of Chapters.

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$\text{CH}_3\text{CHO}$ ,  $\text{CH}_3\text{CH}_2\text{OH}$ ,  $\text{CH}_3\text{OCH}_3$ ,  $\text{CH}_3\text{CH}_2\text{CH}_3$  Ans: The order is :  $\text{CH}_3\text{CH}_2\text{CH}_3 < \text{CH}_3\text{OCH}_3 < \text{CH}_3\text{CHO} < \text{CH}_3\text{CH}_2\text{OH}$  All these compounds have comparable molecular masses  $\text{CH}_3\text{CH}_2\text{OH}$  undergoes extensive intermolecular H-bonding and thus its b.pt. is the highest.

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