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Solution Preview. 1)
Which of the following statements is correct for an electron that has the quantum numbers $n = 4$ and $m_l = -2$? A.

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the electron may be in
a p orbital B. the
electron may be in a d
orbital C. the electron
is in the second
principal shell D. the
electron must have a
spin quantum number
 $m_s = + \frac{1}{2}$

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vapor pressure of liquid
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